

shielding-Effekt der 1α -OH-Gruppe. Bei **2** und **5** handelt es sich um Derivate des Rupicolins A (**6**) [2].

EXPERIMENTELLES

IR: Beckman IR 9, CCl_4 ; $^1\text{H-NMR}$: Bruker WH 270, δ -Werte, TMS als innerer Standard; MS: Varian MAT 711, 70 eV, Direkteinlaß; optische Rotation: Perkin-Elmer-Polarimeter, CHCl_3 . 30 g lufttrockene oberirdische Teile (Dr. S. B. Jones, Univ. of Georgia, Dept. of Botany, Herbar Nr. 76-159) wurden grob zerkleinert mit Et_2O -Petrol (1:2) extrahiert, der Extrakt durch SC (Si gel, Akt.-St. II) und anschließend weiter durch DC (Si gel, GF 254) getrennt. Als Laufmittel dienten Petrol- Et_2O -Gemische bzw. EtOAc . Man erhielt 15 mg **1**, 35 mg **2** (Et_2O -Petrol, 5:1) und 4 mg **5** (Et_2O).

3 α ,4 α -Diacetoxy-3,4-dihydro-rupicolin A-(2- α -acetoxyethylacrylat) (2). Zähes, farbloses Öl, IR: γ -Lacton 1775; OAc 1750, 1230; $\text{C}=\text{C CO}_2\text{R}$ 1720, 1660 cm^{-1} . MS: M^+ m/e , —AcOH 444 (2); 444 — RCO_2H 286 (6); 286 —AcOH 226.099 (75) (ber. für $\text{C}_{15}\text{H}_{14}\text{O}_2$ 226.099).

$$[\alpha]_{24}^{\text{D}} = \begin{array}{cccc} 589 & 578 & 546 & 436 \text{ nm} \\ -175 & -183 & -210 & -382^\circ \end{array} (c = 2.3)$$

10 mg **2** in 1 ml MeOH rührte man 10 min. bei RT mit 25 mg K_2CO_3 in 0.15 ml H_2O . Nach Zugabe von Wasser nahm man CH_2Cl_2 auf und reinigte durch DC (EtOAc). Man erhielt 6 mg **3**, das in 1 ml Aceton mit 5 mg p -Toluolsulfonsäure 21 hr bei RT stehengelassen wurde. Nach Zugabe von Wasser nahm man in CH_2Cl_2 auf und reinigte durch DC (EtOAc). Man erhielt 5 mg **4**, farbloses Öl, MS: m/e ; — CH_3 435.202 (3%) (ber. für $\text{C}_{23}\text{H}_{31}\text{O}_8$ 435.202).

15-Acetoxy-1 β -hydroxy-rupicolin A-(2- α -acetoxyethylacrylat) (5). Zähes, farbloses Öl, IR: OH 3620, 3490; γ -Lacton 1775; OAc 1745, 1235; $\text{C}=\text{C CO}_2\text{R}$ 1725, 1650 cm^{-1} . MS: M^+ m/e 460 (0.1%) —AcOH 400.152 (7) (ber. für $\text{C}_{22}\text{H}_{24}\text{O}_7$ 400.152); — RCO_2H 242 (37); AcOCH (Me) C ($=\text{CH}_2$) CO^+ 141 (100); 141 —AcOH 81 (71).

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LITERATUR

1. Bohlmann, F. und Zdero, C. (1977) *Chem. Ber.* **110**, 1755.
2. Irwin, M. und Geissman, T. (1973) *Phytochemistry* **12**, 863.

Phytochemistry, 1978, Vol. 17, pp. 570–571 Pergamon Press. Printed in England

NEW GERMACRANOLIDES FROM *DICOMA ANOMALA**

FERDINAND BOHLMANN and NGO LE VAN

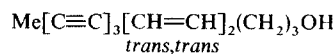
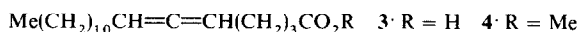
Institute of Organic Chemistry, Technical University, Straße des 17. Juni 135, D 1000 Berlin 12, W. Germany

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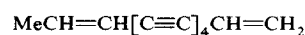
Key Word Index—*Dicoma anomala*; *D. argyrophylla*; Mutisieae; Compositae; new germacranolides.

Abstract—The aerial parts of *D. anomala* contain, besides known compounds, two new 7,8-germacranolides. The structures have been elucidated by spectroscopic methods. *D. argyrophylla* lacks lactones.

The South African genus *Dicoma* has been very little investigated chemically. *D. zeyheri* afforded the acetylenic compounds **1** and **2**, also present in species from the tribe Centaureae, and the allenic acid **3** together with the ester **4** [1]. From *D. anomala* Sond. caffeic acid only has been reported [2]. The roots of *D. anomala* Sond. ssp. *cirsoides* (Harv.) Willd. have now afforded the triynediene **5** [3], while the aerial parts contain **6** [3] and lupeol (**7**) together with some sesquiterpene lactones. The most polar is the known lactone albicolide (**8**) [4], while the main constituent with molecular formula $\text{C}_{19}\text{H}_{24}\text{O}_7$ must be a diacetate with an additional free hydroxyl group. The $^1\text{H-NMR}$ -data (Table 1) clearly show that the acetate groups are located at C-14 and C-15 and that the remaining O-functions are at C-6 and C-8. To decide



5



6

	9	10	11	12
R	CH_2OAc	CH_2OAc	CHO	CHO
R'	H	Ac	H	Ac

* Part 134 in the series 'Naturally Occurring Terpene Derivatives'; for part 133 see: Bohlmann, F., Brindöpke, G. and Rastogi, R. C. (1978) *Phytochemistry* **17**, 475

Table 1. $^1\text{H-NMR}$ -data of **9**, **10** and **12**

H	9 C_6D_6 77°	10 C_6D_6 77°	12 CDCl_3
1	<i>dd</i> (br) 5.30	<i>dd</i> (br) 5.28	<i>dd</i> 6.60
2	<i>m</i> 1.96	<i>m</i> 1.9	$\left\{ \begin{array}{l} m \text{ 2.65} \\ m \text{ 2.44} \end{array} \right.$
3	<i>ddd</i> 2.04	<i>m</i> 2.00	$\left\{ \begin{array}{l} m \text{ 2.40} \end{array} \right.$
3'	<i>m</i> 1.82	<i>m</i> 1.8	
5	<i>d</i> (br) 5.13	<i>d</i> (br) 5.02	<i>d</i> (br) 5.09
6 β	<i>dd</i> 4.33	<i>dd</i> 5.45	<i>d</i> 5.24
7 α	<i>dddd</i> 2.51	<i>dddd</i> 2.69	<i>m</i> 2.78
8 β	<i>ddd</i> 3.45	<i>ddd</i> 3.70	<i>m</i> 4.09
9 α	<i>dd</i> 2.36	<i>dd</i> 2.35	$\left\{ \begin{array}{l} m \text{ 2.40} \end{array} \right.$
9 β	<i>dd</i> 2.20	<i>dd</i> 2.25	
13	<i>dd</i> 6.31	<i>d</i> (br) 6.20	<i>d</i> 6.34
13'	<i>dd</i> 6.10	<i>d</i> (br) 5.51	<i>d</i> 5.82
14	<i>d</i> (br) 4.60	<i>d</i> (br) 4.68	$\left\{ \begin{array}{l} s \text{ 9.53} \end{array} \right.$
14'	<i>d</i> (br) 4.44	<i>d</i> (br) 4.44	
15	<i>d</i> 4.55	<i>d</i> 4.58	<i>d</i> 4.78
15'	<i>d</i> 4.08	<i>d</i> 4.52	<i>d</i> 4.68
OAc	<i>s</i> 1.84	<i>s</i> 1.84	<i>s</i> 2.10
	<i>s</i> 1.73	<i>s</i> 1.74	<i>s</i> 1.09

$J(\text{Hz})$: 1,2 = 7; 1,2' = 8; 2,3 = 3.5; 3,3' = 12; 5,6 = 13; 6,7 = 9; 7,8 = 8; 7,13 = 3; 8,9 α = 3; 8,9 β = 5; 9 α ,9 β = 15; 13,13' = 1; 14,14' = 13; 15,15' = 12.

whether a 6,7- or a 7,8-lactone is present, we acetylated the alcohol. In the NMR spectrum of the triacetate, there was a considerable downfield shift for the proton under what was previously the hydroxyl group. At room temperature the spectra are not completely clear; however in deuteriobenzene at 77° all signals can be interpreted (see Table 1). As this proton is also coupled with the olefinic proton at C-5 the structure must be **9**, with **10** being the triacetate. The 14-desacetoxy compound we have named dicomanolide. In a second lactone, which could only be purified after acetylation, the situation at C-14 must be changed to an aldehyde. The position of this group follows from the observed shift of the signal of 1-H. Therefore the structure is **11**.

The roots of *D. argyrophylla* Oliv. contain **1-4** and **15**, while the aerial parts only afford linoleic and linolenic acid together with their methyl esters. Further species must be investigated to see whether the lactones of *D. anomala* constitute an exception or not.

EXPERIMENTAL

IR: Beckmann IR9, CCl_4 ; $^1\text{H-NMR}$: Bruker WH 270, δ -values, TMS as int. stand., signal interpretation established by double resonance experiments. MS: Varian MAT 711, 70 eV; optical rotation: Perkin-Elmer-polarimeter, CHCl_3 . The air

dried plant material, collected in February 1977 in Natal, was extracted with Et_2O -petrol (1:2) and the resulting extracts were separated first by column chromatography (Si gel, act. grade II) and further by TLC (Si gel GF 254) using Et_2O -petrol and Et_2O -MeOH as solvents. Known compounds were identified by comparison of the IR and NMR spectra.

Dicoma anomala Sond. ssp. *cirsioides* (Harv.) Willd. (Voucher 77/53). 160 g roots afforded 1 mg **5** and 72 g aerial parts 0.1 mg **6**, 40 g **7**, 170 mg **9** (ether) 10 mg **8** and 20 mg **11** (isolated as diacetate).

Dicoma argyrophylla Oliv. (Voucher 77/172). 230 g roots afforded 1 mg **13**, 1 mg **1**, 10 mg **4**, 110 mg **3** and 1 mg **2**, while 360 g aerial parts yielded 40 mg linoleic and linolenic acid methyl ester and 200 mg of the corresponding acids (a 1:1).

14-Acetoxydicomanolide (**9**). Colourless oil, IR: OH 3620; lactone 1775; OAc 1745, 1220 cm^{-1} . MS: M^+ m/e 364.152 (0.5%) (calc. for $\text{C}_{19}\text{H}_{24}\text{O}_7$ 364.152); $-\text{OAc}$ 305 (25); $-\text{AcOH}$ 304 (3); 304 $-\text{H}_2\text{C}=\text{C}=\text{O}$ 262 (6); 262 $-\text{H}_2\text{O}$ 244 (26); MeCO^+ 43 (100).

$$[\alpha]_{24}^{\lambda} = \frac{589}{-35.3} \frac{578}{-37} \frac{546}{-43.7} \frac{436}{-89.9} \text{ nm} (c = 2.6)$$

To 30 mg **9** in 2 ml Ac_2O and 0.1 ml Py 10 mg 4-pyrrolidino-pyridine [5] was added. After 12 hr the product was purified after usual work up by TLC (Et_2O -petrol, 1:1, $\times 3$). 20 mg **10** were obtained, colourless oil, IR: methylene lactone 1775; OAc 1740, 1230 cm^{-1} . MS: M^+ m/e ; $-\text{OAc}$ 347.149 (5%) (calc. for $\text{C}_{19}\text{H}_{23}\text{O}_6$ 347.149); $-\text{AcOH}$ 346 (1); 346 $-\text{OAc}$ 287 (2); Ac^+ 43 (100).

$$[\alpha]_{24}^{\lambda} = \frac{589}{+15} \frac{578}{+15.6} \frac{546}{+17.5} \frac{436}{+27.4} \text{ nm} (c = 1.06)$$

14-Oxodicomanolide (**11**). Not isolated in pure state, after acetylation as above and TLC (Et_2O , $\times 4$) 20 mg **12** were obtained, colourless oil, IR: $\text{C}=\text{CHO}$ 2720, 1695, 1645; lactone 1775; OAc 1750, 1235 cm^{-1} . MS: M^+ m/e 362 (1%); $2 \times \text{AcOH}$ 242.093 (100) (calc. for $\text{C}_{15}\text{H}_{14}\text{O}_3$ 242.094).

$$[\alpha]_{24}^{\lambda} = \frac{589}{+68} \frac{578}{+71} \frac{564}{+81} \frac{436}{+137} \text{ nm} (c = 1.9)$$

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REFERENCES

- Bohlmann, F., Rode, K. M. and Grenz, M. (1967) *Chem. Ber.* **100**, 3201.
- Karrer, W. (1958) *Konstitution und Vorkommen von organischen Pflanzenstoffen*. Birkhäuser-Verlag, Basel.
- Bohlmann, F., Burkhardt, T. and Zdero, C. (1973) *Naturally Occurring Acetylenes*. Academic Press, London.
- Suchy, M., Samek, Z., Herout, V. and Sorm, F. (1967) *Coll. Czech. Chem. Commun.* **32**, 3934.
- Steglich, W. and Höfle, G. (1965) *Angew. Chem.* **81**, 1001.

